



# Premio ambiente & futuro 2014

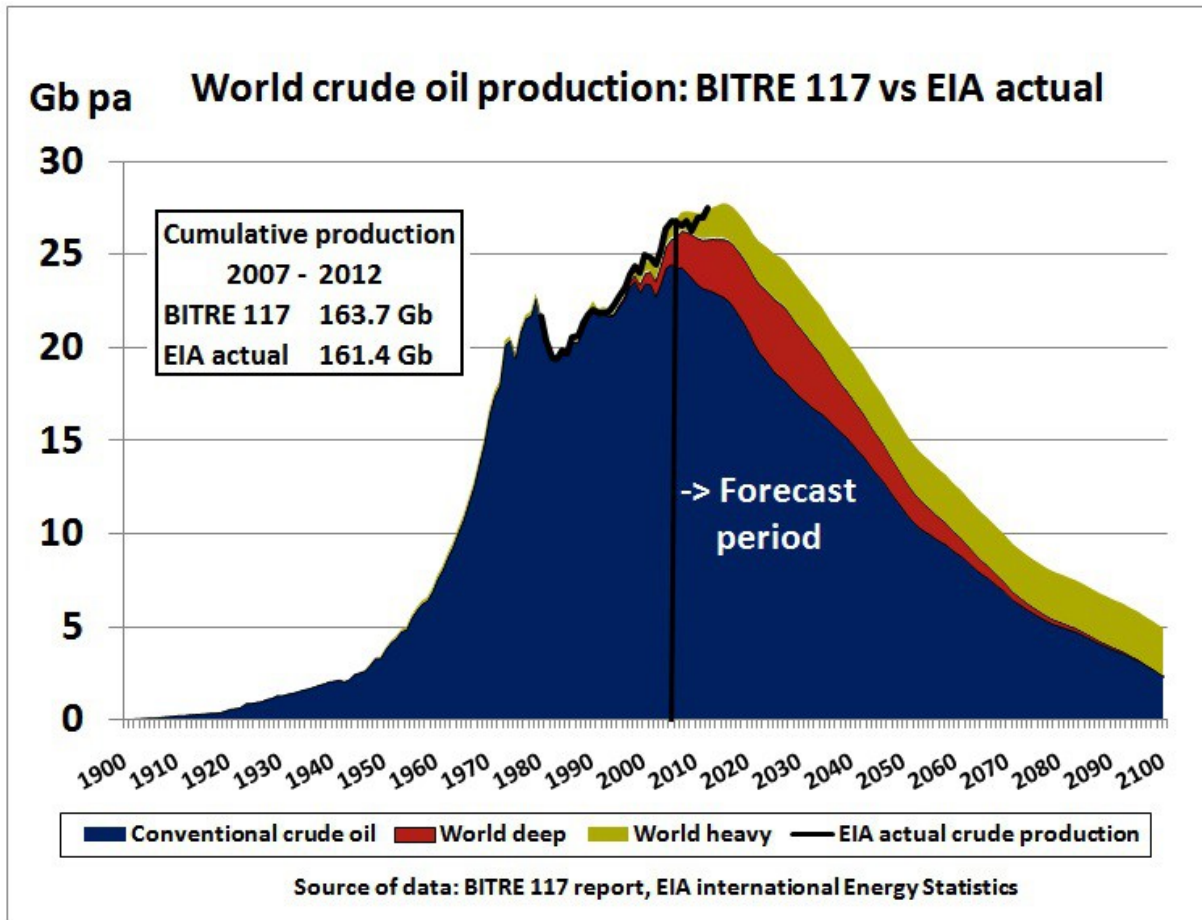
Emilio Tagliavini

Fonti rinnovabili per la chimica e per  
l'energia: un'opportunità sostenibile  
che non possiamo rifiutare

*Ravenna – Sala Cavalcoli, 1 Dicembre 2014*

# Perché accettare la sfida delle fonti rinnovabili per la Chimica e per l'Energia?

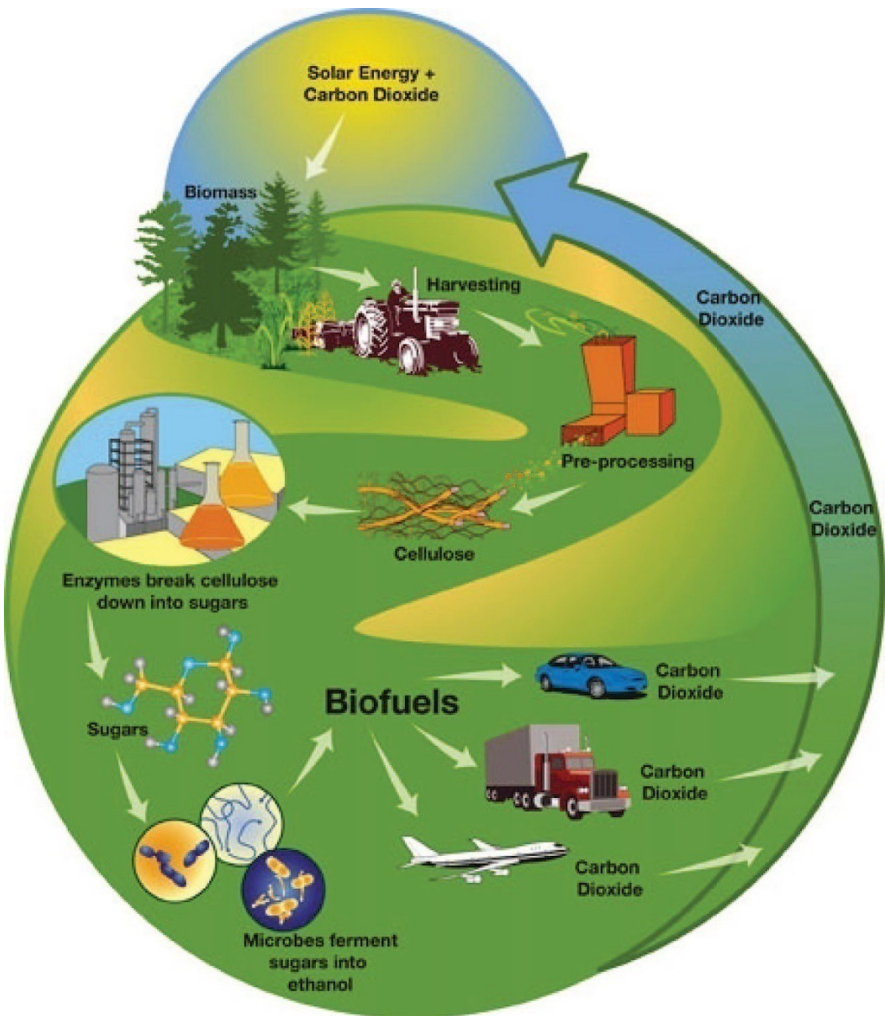
# Fonti Rinnovabili - Sostenibilità



Entro poco tempo la quantità di materia organica da fonti fossili disponibile per la Chimica e per l'Energia è destinata a diminuire

# Fonti Rinnovabili

## Riduzione dell'Effetto Serra

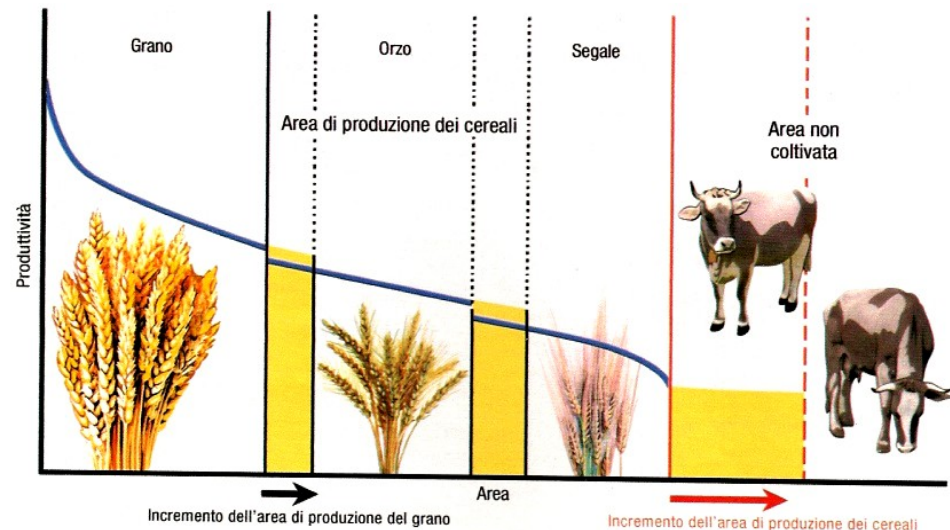


*Angew. Chem. Int. Ed.* 2008, 47, 9200 – 9211

# Uso del Suolo: l'Effetto ILUC

L'uso indiretto del suolo (Indirect Land Use Change - ILUC) consiste nella destinazione di aree sempre meno produttive all'agricoltura che richiede superfici sempre più vaste

**ATTENZIONE!**  
**Occorre anche considerare l'uso indiretto del suolo**



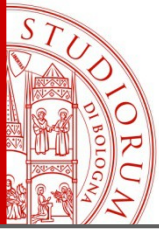
# La Crisi dei Prezzi 2007-2008

**Evoluzione dei prezzi di alcuni prodotti agricoli (commodity) nel mercato internazionale**

Prodotto	Gennaio 2006 (US \$)	Febbraio 2008 (US \$)	% di incremento
Mais (FOB) (Chicago)	120	250	
Frumento (FOB) (Chicago)	170	439	158 %
Olio di soia (FOB) (Amsterdam)	532	1.400	163 %
Pellets di soia (CIF) (Rotterdam)	207	453	118 %
Latte in polvere (FOB) (Oceania)	2.175	4.550	109 %
Carne congelata (FOB) (Argentina)	1.965	2.773	41 %

Fonte: FAO - ICO



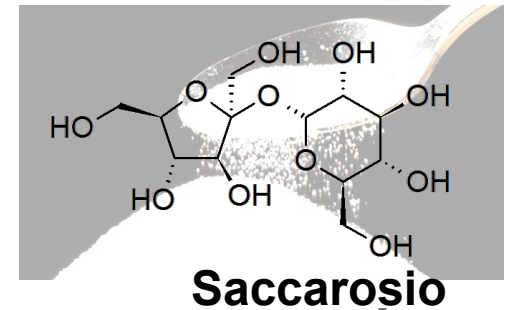
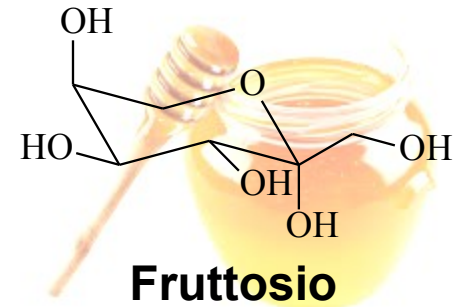
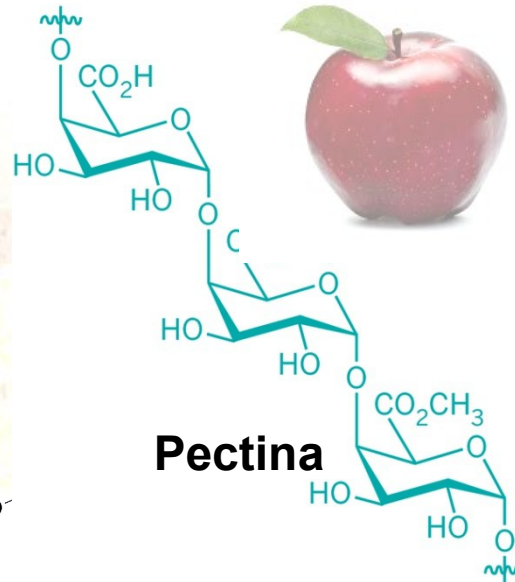
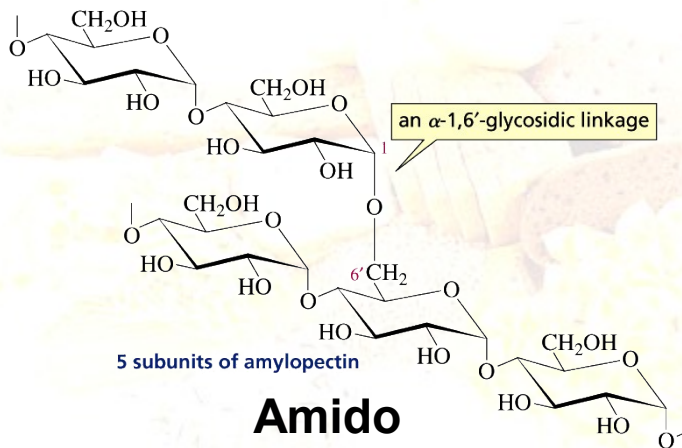
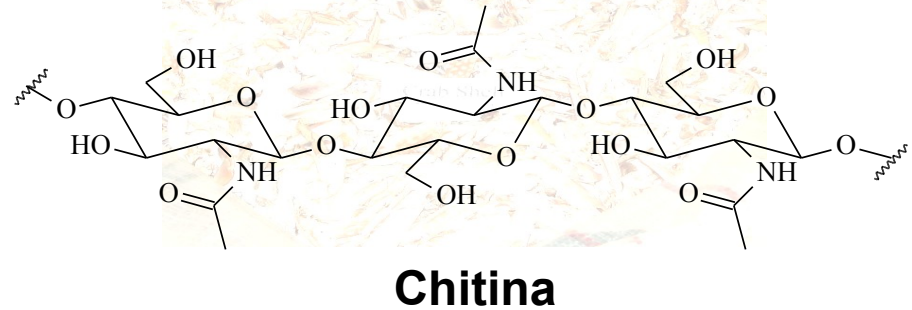
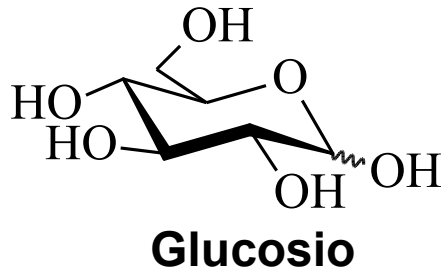


# Fonti Rinnovabili

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Cosa sono?  
**BIOMASSA**

# Carboidrati





# Trigliceridi e Polioli

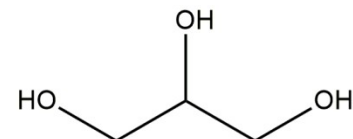
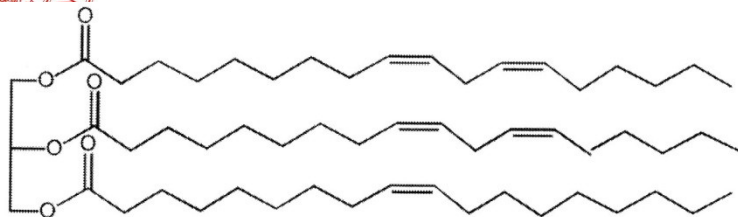


Table 3.1 Common naturally occurring fatty acids

### Saturated

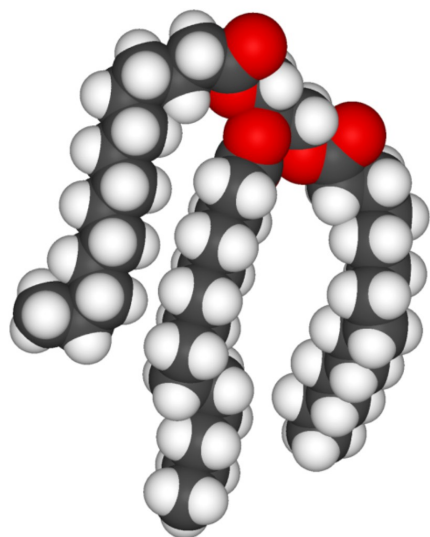
butyric	$\text{CH}_3(\text{CH}_2)_2\text{CO}_2\text{H}$	4:0	stearic	$\text{CH}_3(\text{CH}_2)_{16}\text{CO}_2\text{H}$	18:0
caproic*	$\text{CH}_3(\text{CH}_2)_4\text{CO}_2\text{H}$	6:0	arachidic	$\text{CH}_3(\text{CH}_2)_{18}\text{CO}_2\text{H}$	20:0
caprylic*	$\text{CH}_3(\text{CH}_2)_6\text{CO}_2\text{H}$	8:0	behenic	$\text{CH}_3(\text{CH}_2)_{20}\text{CO}_2\text{H}$	22:0
capric*	$\text{CH}_3(\text{CH}_2)_8\text{CO}_2\text{H}$	10:0	lignoceric	$\text{CH}_3(\text{CH}_2)_{22}\text{CO}_2\text{H}$	24:0
lauric	$\text{CH}_3(\text{CH}_2)_{10}\text{CO}_2\text{H}$	12:0	cerotic	$\text{CH}_3(\text{CH}_2)_{24}\text{CO}_2\text{H}$	26:0
myristic	$\text{CH}_3(\text{CH}_2)_{12}\text{CO}_2\text{H}$	14:0	montanic	$\text{CH}_3(\text{CH}_2)_{26}\text{CO}_2\text{H}$	28:0
palmitic	$\text{CH}_3(\text{CH}_2)_{14}\text{CO}_2\text{H}$	16:0	melissic	$\text{CH}_3(\text{CH}_2)_{28}\text{CO}_2\text{H}$	30:0

\*To avoid confusion, systematic nomenclature (hexanoic, octanoic, decanoic) is recommended

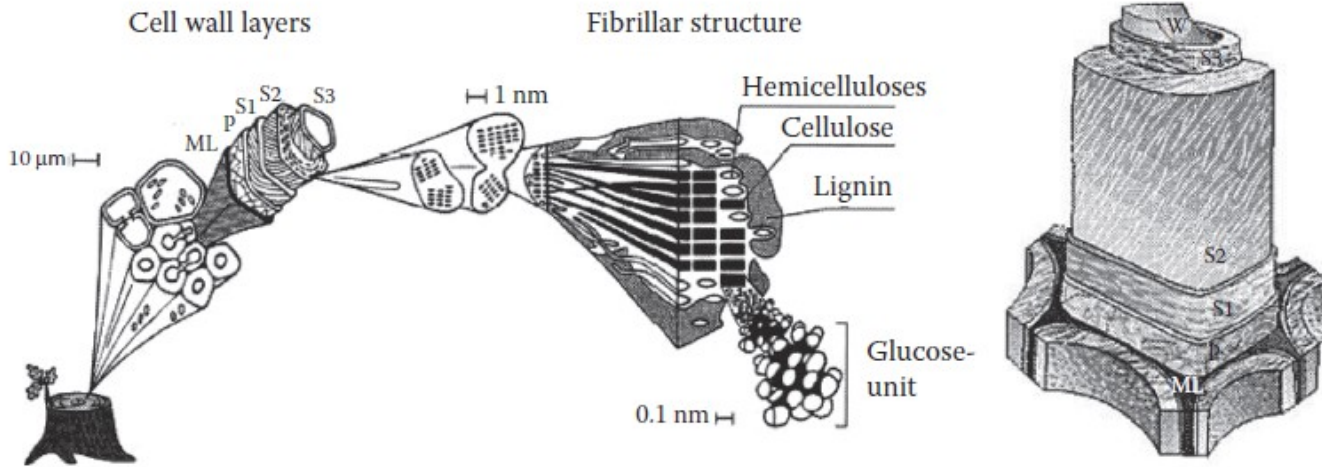
### Unsaturated

palmitoleic	$\text{CH}_3(\text{CH}_2)_5\text{CH}=\text{CH}(\text{CH}_2)_7\text{CO}_2\text{H}$	16:1 (9c)
oleic	$\text{CH}_3(\text{CH}_2)_7\text{CH}=\text{CH}(\text{CH}_2)_7\text{CO}_2\text{H}$	18:1 (9c)
cis-vaccenic	$\text{CH}_3(\text{CH}_2)_5\text{CH}=\text{CH}(\text{CH}_2)_9\text{CO}_2\text{H}$	18:1 (11c)
linoleic	$\text{CH}_3(\text{CH}_2)_4\text{CH}=\text{CHCH}_2\text{CH}=\text{CH}(\text{CH}_2)_7\text{CO}_2\text{H}$	18:2 (9c, 12c)
$\alpha$ -linolenic	$\text{CH}_3\text{CH}_2\text{CH}=\text{CHCH}_2\text{CH}=\text{CHCH}_2\text{CH}=\text{CH}(\text{CH}_2)_7\text{CO}_2\text{H}$	18:3 (9c, 12c, 15c)
$\gamma$ -linolenic	$\text{CH}_3(\text{CH}_2)_4\text{CH}=\text{CHCH}_2\text{CH}=\text{CHCH}_2\text{CH}=\text{CH}(\text{CH}_2)_4\text{CO}_2\text{H}$	18:3 (6c, 9c, 12c)
gadoleic	$\text{CH}_3(\text{CH}_2)_9\text{CH}=\text{CH}(\text{CH}_2)_7\text{CO}_2\text{H}$	20:1 (9c)
arachidonic	$\text{CH}_3(\text{CH}_2)_4\text{CH}=\text{CHCH}_2\text{CH}=\text{CHCH}_2\text{CH}=\text{CHCH}_2\text{CH}=\text{CH}(\text{CH}_2)_3\text{CO}_2\text{H}$	20:4 (5c, 8c, 11c, 14c)
eicosapentaenoic (EPA)	$\text{CH}_3\text{CH}_2\text{CH}=\text{CHCH}_2\text{CH}=\text{CHCH}_2\text{CH}=\text{CHCH}_2\text{CH}=\text{CH}(\text{CH}_2)_3\text{CO}_2\text{H}$	20:5 (5c, 8c, 11c, 14c, 17c)
erucic	$\text{CH}_3(\text{CH}_2)_7\text{CH}=\text{CH}(\text{CH}_2)_{11}\text{CO}_2\text{H}$	22:1 (13c)
docosapentaenoic (DPA)	$\text{CH}_3\text{CH}_2\text{CH}=\text{CHCH}_2\text{CH}=\text{CHCH}_2\text{CH}=\text{CHCH}_2\text{CH}=\text{CHCH}_2\text{CH}=\text{CH}(\text{CH}_2)_5\text{CO}_2\text{H}$	22:5 (7c, 10c, 13c, 16c, 19c)
docosahexaenoic (DHA)	$\text{CH}_3\text{CH}_2\text{CH}=\text{CHCH}_2\text{CH}=\text{CHCH}_2\text{CH}=\text{CHCH}_2\text{CH}=\text{CHCH}_2\text{CH}=\text{CH}(\text{CH}_2)_2\text{CO}_2\text{H}$	22:6 (4c, 7c, 10c, 13c, 16c, 19c)
nervonic	$\text{CH}_3(\text{CH}_2)_7\text{CH}=\text{CH}(\text{CH}_2)_{13}\text{CO}_2\text{H}$	24:1 (15c)

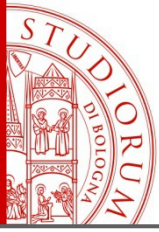
all double bonds are Z (cis)



# Lignocellulosa



La Biomassa Lignocellulosica è il materiale organico più abbondante sulla terra. E' prodotto il quantità pari a  $170 \times 10^9$  ton/anno (Bozell and Patel, 2006), dei quali solo il 3% circa è raccolto ed utilizzato come cibo o per altri scopi.



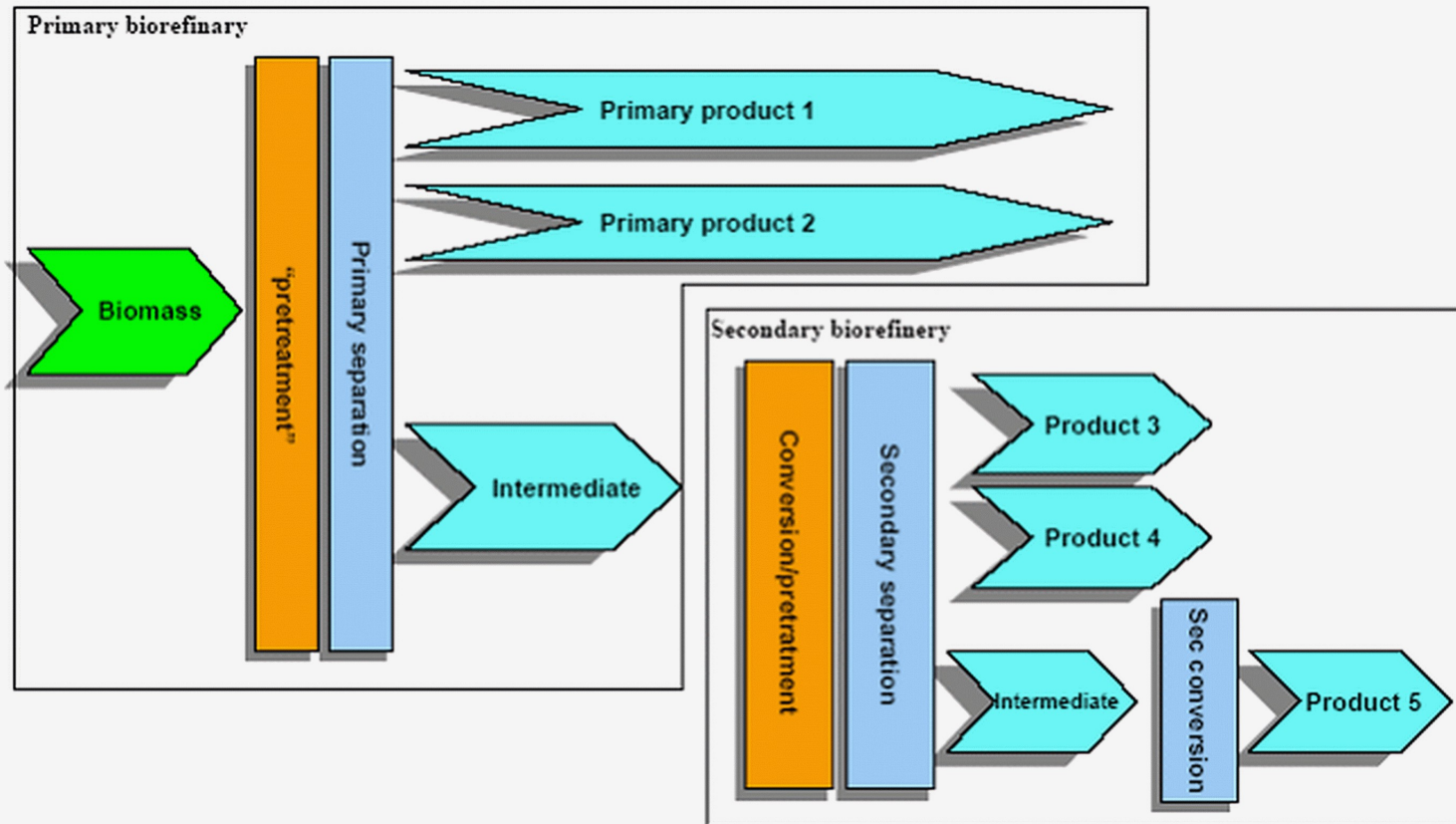
# Fonti Rinnovabili

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Come usarle?  
**BIORAFFINERIA**

# Bioraffinerie

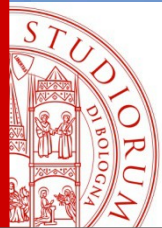
## Biorefineries





# Generazioni di Bioraffinerie

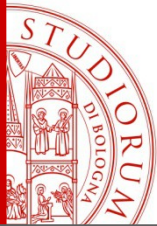
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# Generazioni di Bioraffinerie

## Biorefineries

<u>Generazione</u>	<u>Materia prima</u>	<u>Prodotti</u>
<u>Bioraffinerie di prima generazione</u>	<u>Zucchero, amido, oli vegetali, grassi animali</u>	<u>Bio-alcoli, biodiesel, biogas, biosyngas. Bio-platform chemicals, biopolimeri</u>
<u>Bioraffinerie di seconda generazione</u>	<u>Coltivazioni non alimentari, paglia, residui agricoli, colture energetiche</u>	<u>Bio-alcoli, biodiesel, biohydrogen, bio-Fischer-Tropsch diesel, Bio-platform chemicals,</u>
<u>Bioraffinerie di terza generazione</u>	<u>Algae</u>	<u>Oli Vegetali, bio-diesel</u>
<u>Bioraffinerie di quarta generazione</u>	<u>Oli vegetali, Rifiuti urbani</u>	<u>Bio-gasolio</u>

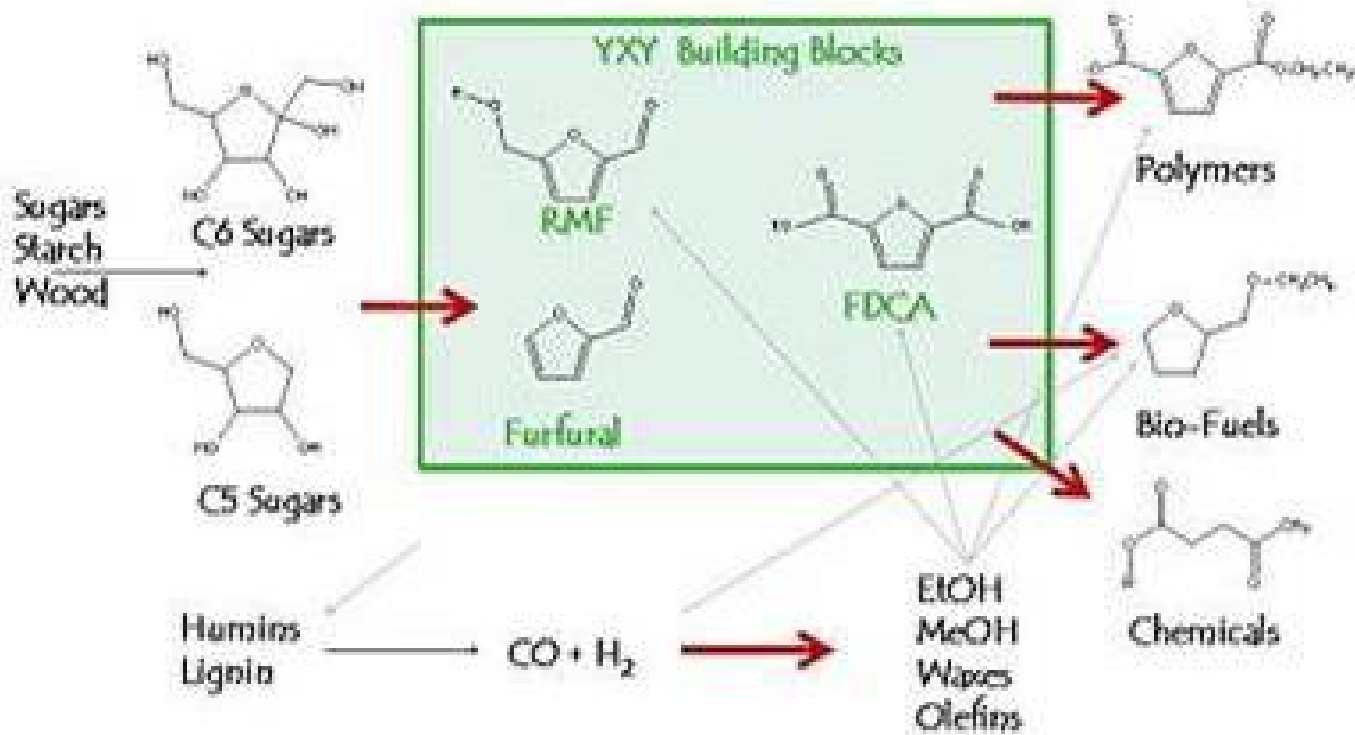


# Fonti Rinnovabili

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**Per la Chimica**  
**Platform Chemicals**  
**Biopolimeri**

# Biorffinaria basata su Idrossimetilfuralaldeide HMF



**Avantium Co.** in Olanda ha prodotto **HMF** dalla Biomassa a costi competitivi. I derivati di HMF sono commercializzati col marchio **YXY Technology**



# Bio-Plastiche Biodegradabili

## Modified Starch Polymers

Si basano su **Amido modificato** o miscele di Amido e altri polimeri biodegradabili

## Mater-Bi® (Νοπαμοντσ) Σολανψλ®

(Ροδενβυργς) Βιοπλαστ® (Βιοτεχ)

Υσι: Ιμβαλλαγι, Πνευματιχι, Ριεμπιτιπο,

Ποσατε υσα ε γεττα.

## Polylactic Acid (PLA)

Polimerizzazione dell'Acido Lattico (via Lattide)

Ottime proprietà meccaniche e/o plasticità. A seconda della composizione stereoisomerica.

Usi: Imballaggi, Fibre tessili, Riempitivo, Ausili medici e Impianti.

## Polyhydroxyalcanoates (PHA)

Materiale di riserva di energia e massa di alcuni Batteri



Ottime proprietà meccaniche.

Candidato a sostituire PET e HDPE.

Usi: Oggetti, Imballaggi, Ausili medici.

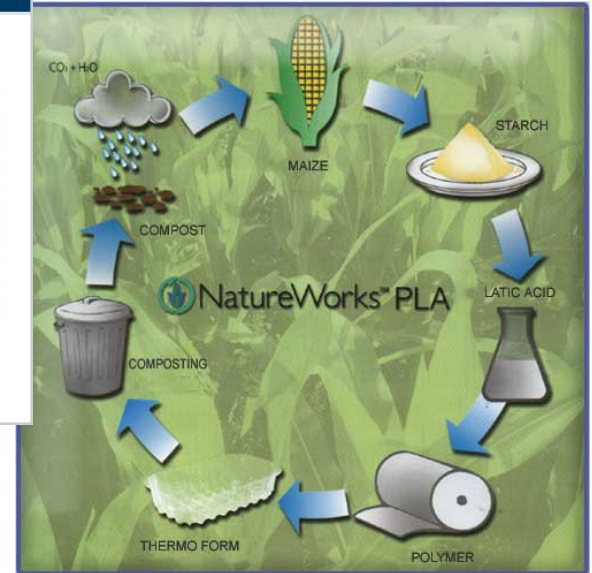
# Bioplastiche



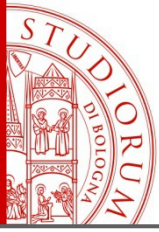
Amido modificato



Poliidrossialcanoati PHA



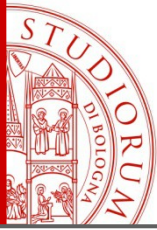
Acido Polilattico PLA



# Fonti Rinnovabili

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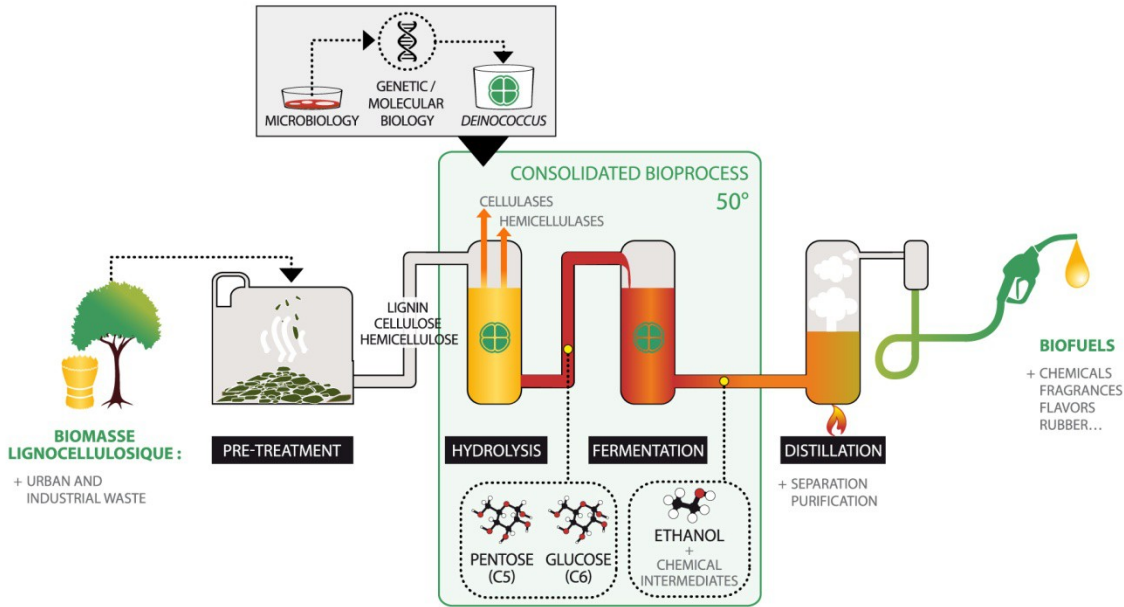
**Per l'Energia**  
**BioEtanolo**  
**BioDiesel**  
**BioGas SynGas**



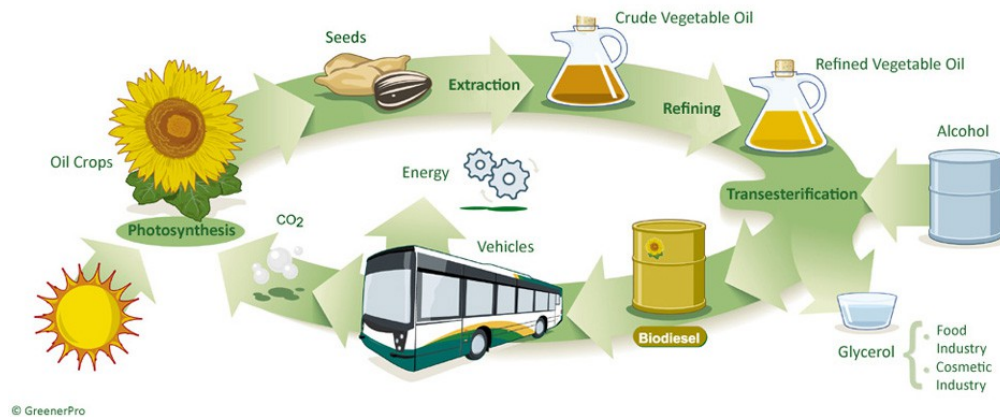
# Come ottenere Biocarburanti dalla Biomassa

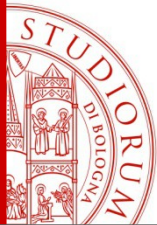
- **Processi Chimico-fisici**
  - Estrazione, separazione (Lipidi da Alghe)
- **Processi Chimici**
  - Transesterificazione, conversione del Syngas
- **Processi Biochimici**
  - Fermentazione (Bioetanolo, Bioldrogeno)
  - Biogas
- **Processi Termochimici**
  - Pirolisi
  - Gassificazione

# I BioCarburanti



## The Biodiesel Cycle





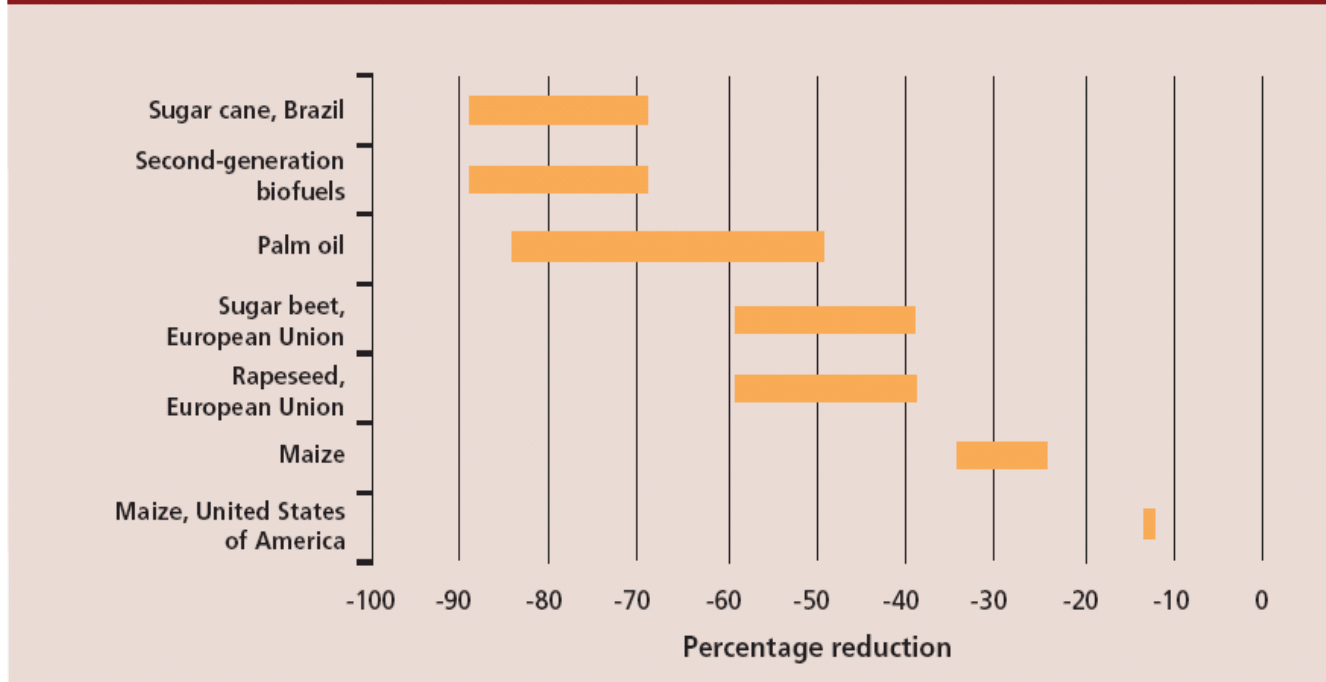
# Perché i BioCarburanti?

- *Rinnovabilità*
  - La riserva di Biomassa si rinnova rapidamente (tipicamente ogni anno)
- *Sostenibilità ambientale*
  - Potenzialmente non c'è produzione netta di gas-serra o, perfino, una riduzione del contenuto di CO2 in atmosfera
- *Economia*
  - L'instabilità dei prezzi dei combustibili fossili danneggia l'economia
- *Aspetti Geopolitici*
  - La produzione di combustibili fossili è concentrata in aree politicamente instabili
  - Prezzi e disponibilità di combustibili fossili sono stati usati come deterrente politico

# Quanto è vero?

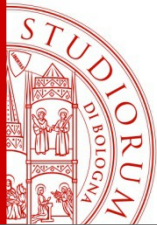
FIGURE 23

Reductions in greenhouse gas emissions of selected biofuels relative to fossil fuels



Note: Excludes the effects of land-use change.

Sources: IEA, 2006, and FAO, 2008d.



ALMA MATER STUDIORUM  
UNIVERSITÀ DI BOLOGNA

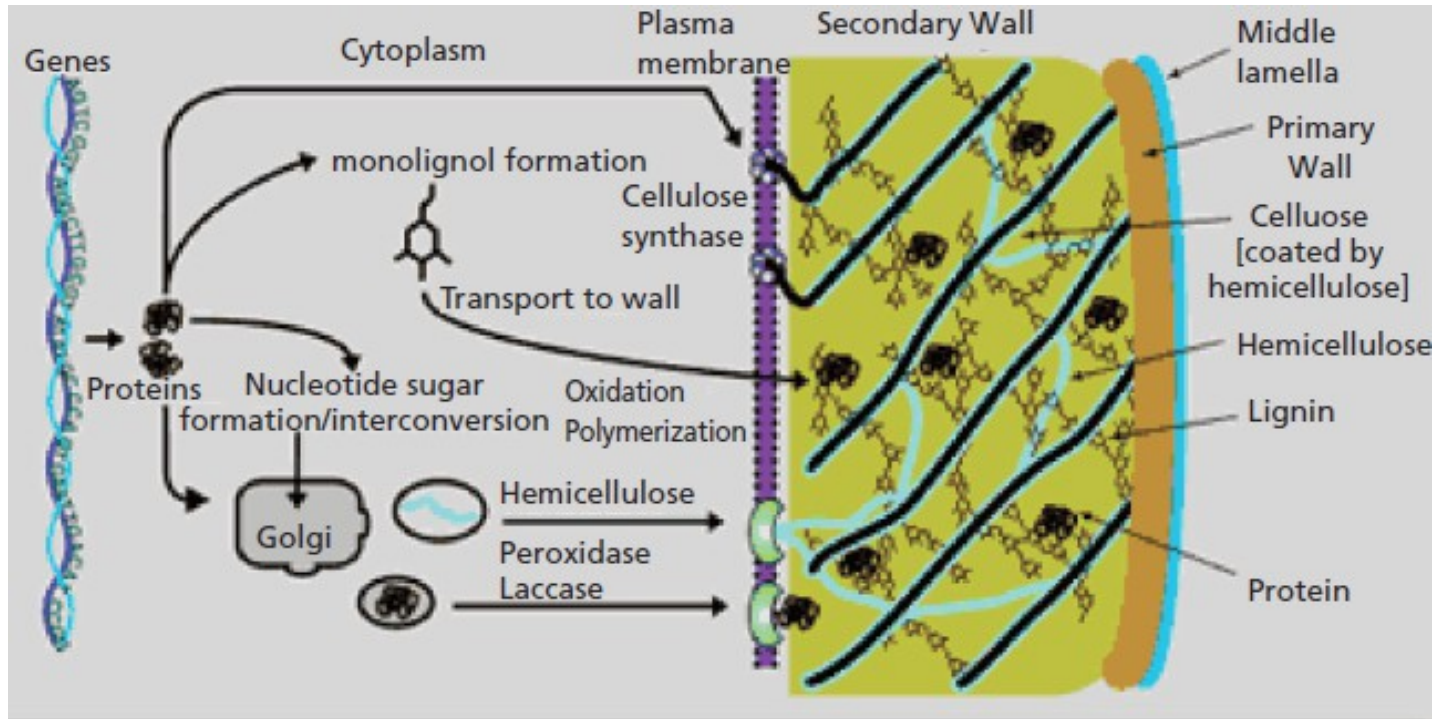
# GRAZIE PER L'ATTENZIONE

*EMILIO TAGLIAVINI DIPARTIMENTO DI CHIMICA, UNIVERSITA' DI BOLOGNA*

*emilio.tagliavini@unibo.it*



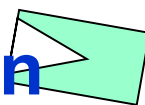
# Biogenesis of Lignocellulose



# La Via Termochimica



**Gasification**



**Gas CO, H<sub>2</sub>  
(Syngas)**

**Pyrolysis  
Fast/flash**



**Liquid  
Bio-oil**



[www.dynamotive.com](http://www.dynamotive.com)

**Carbonization**

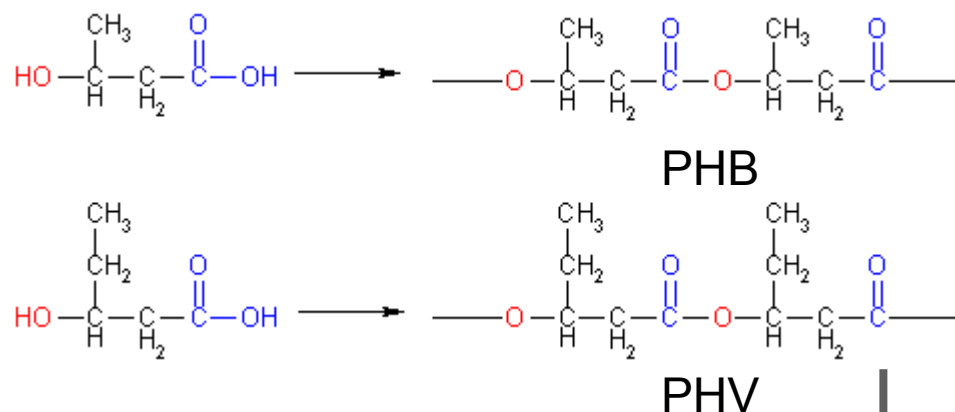
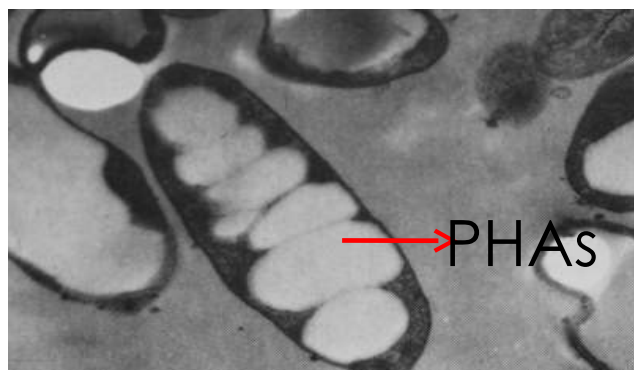


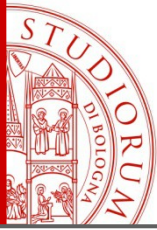
**Solid (*char*)**



# Polyhydroxyalkanoates PHA

- Produced by Micoorganisms
- Pure Cultures, Mixed cultures, Plants
- Eubacteria and Archaea; 75 different genera of Gram+ and Gram-bacteria ( $\approx$  300 bacteria)
- Insoluble inclusions in the cytoplasm
- Synthesized and stored inside the cells as energy and carbon storage material
- 90% of PHA by cell dry weight, usually under conditions of nutrient limitation (N, P, Mg) with excess of carbon



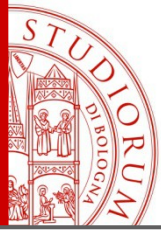


# Starch-based Bioplastics

- So-called **Plastified Starch** consists of starch composites with *starch content being more than 50% by mass*. Exhibits mechanical properties similar to conventional plastics but at a lower price.
- Bioplastics with amount of *starch higher than 50%* are usually referred to as **Thermoplastic Starch**.
- Usually, the components to blend with starch are aliphatic polyesters, **polyvinyl alcohol (PVA)** and biopolymers. The

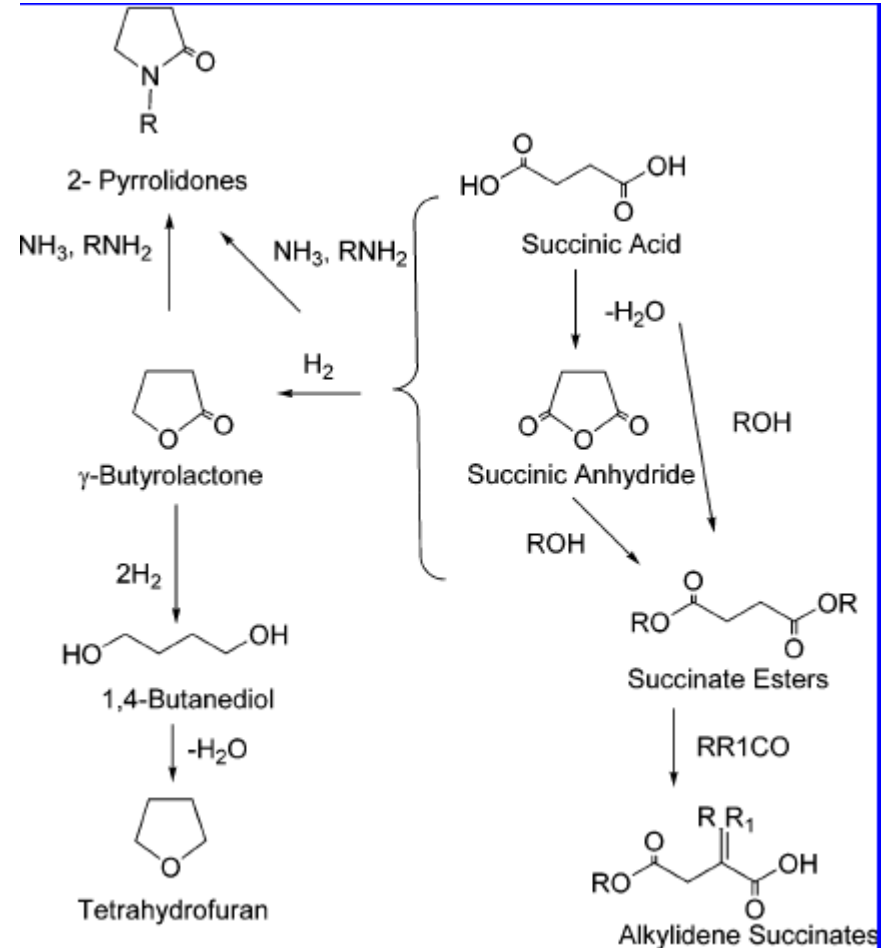
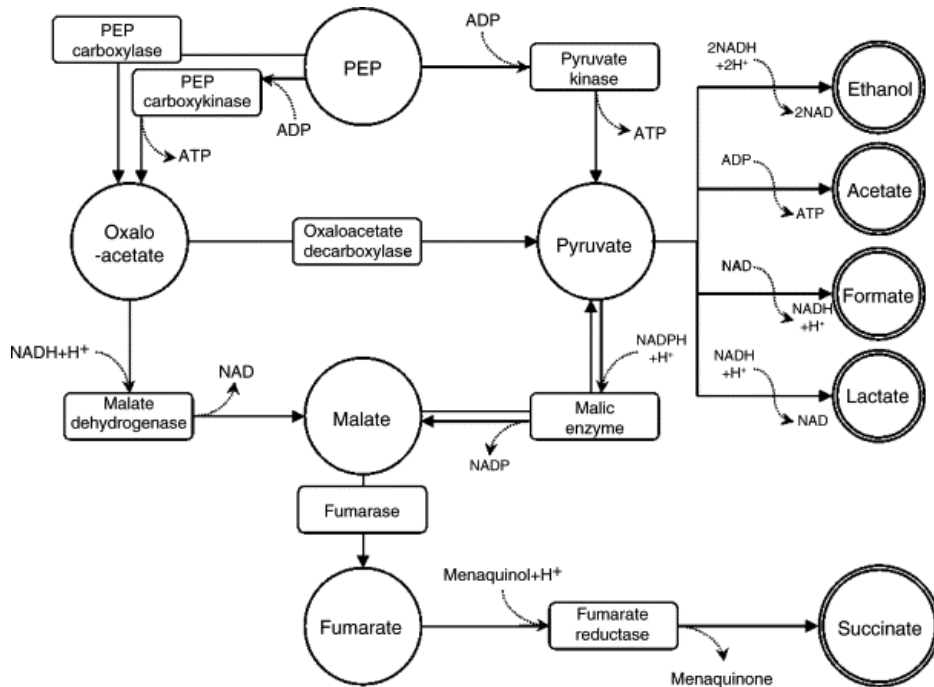
# Polyhydroxyalkanoates PHA

- The most common representative of these biopolymers is poly-3-hydroxybutyrate (PHB). PHB constitutes a rather stiff and brittle material with a low extension to break. These mechanical properties can be enormously enhanced by incorporation of different co-monomers into the PHB matrix. This results in co- and terpolyesters possessing mechanical properties similar to high-tech materials.



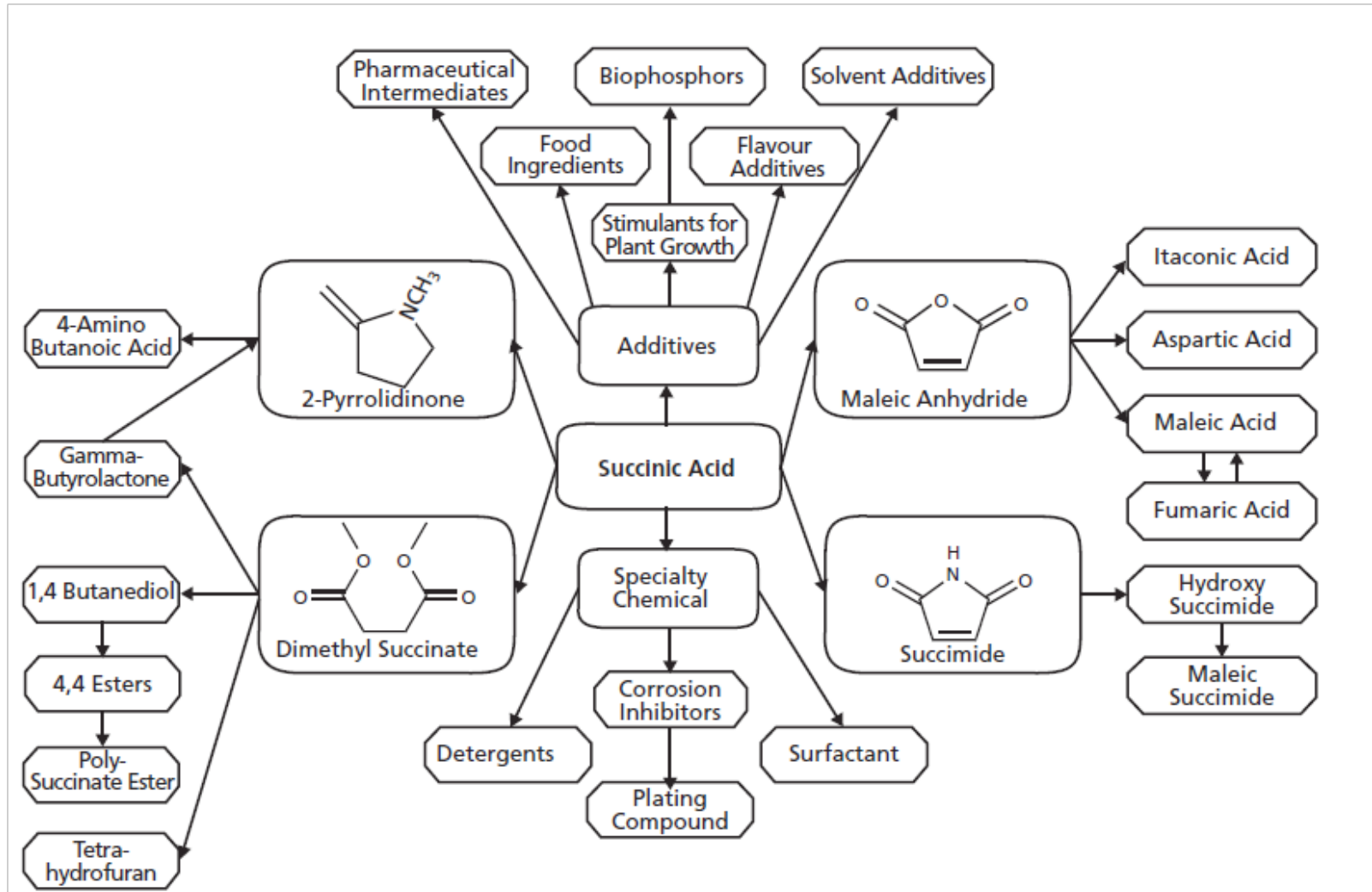
# Succinic Acid Platform

From Sugars *via* Phosphoenolpyruvate (PEP) carboxylation to give Oxalacetic Acid, followed by the reaction of Krebs Cycle affording Fumaric Acid that is finally reduced to Succinic Acid

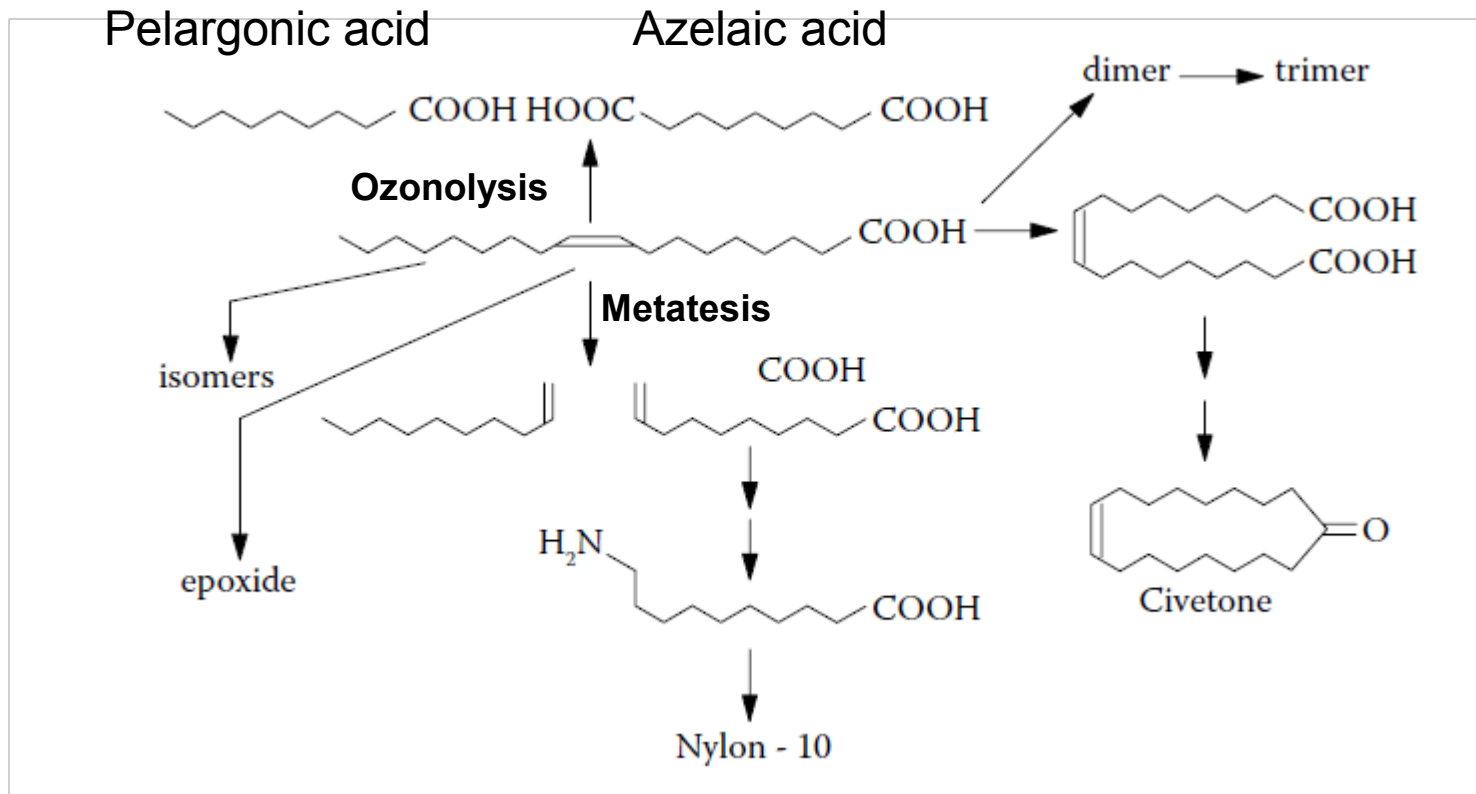


BioAmber Inc. has realized the industrial implementation exploiting genetically engineered *E. coli* strain

# Chemicals from Succinic Acid



# Oleic Acid Platform



EVONIC in Germany ha developed a biobased production of Nylon 10-10



# Third generation: Biofuels from

Algae



- Fast growth
- Very efficient photosynthesis: better utilization of sunlight energy
- High lipid content: some strain contain up to 30% weight lipids (on dry mass)
- Little competition with food-chain use
- Use of marginal or desertic lands
- Marine and freshwater species available
- Main drawback: need of dewatering. It is highly energy costing.